

Research Article

Comparative Destructive Effect of Waterborne Zinc Nanoparticles and Zinc sulfate on *Capoeta capoeta gracilis* Hematological Indices

Zohre Soltani¹, Rasool Ghorbani¹, Seyed Aliakbar Hedayati¹, Hamed Ghafari Farsani², Mohammad Hasan Gerami^{3*}

¹Department of Fisheries, Gorgan University of Agricultural and Natural Resources, Gorgan, Iran

²Young Researchers and Elite Club, Shahrekord Branch, Islamic Azad University, Shahrekord, Iran

³Young Researchers and Elite Club, Shiraz Branch, Islamic Azad University, Shiraz, Iran

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Abstract:

This study aimed to investigate the disruptive effect of Zinc nanoparticles and Zinc sulfate on the hematological indices of *Capoeta capoeta gracilis* and comparing the severity of two forms of this metal. For this purpose, 70 species of *C. capoeta gracilis* with 15 ± 3.5 gr mean weight, were exposed for 14 days to sublethal toxicity of 5, 20 and 200 ppm toxicant concentration and one control group (no toxicant), both for zinc sulfate and zinc nanoparticles treatments with three replicate. There were no significant differences between aquariums in water quality and the following were constant: pH: 7.56 ± 0.45 (TS1); temperature: $19 \pm 1^\circ\text{C}$; hardness: 293 ± 2.35 ppm and dissolved oxygen: 7.80 ± 1 mg L⁻¹. Blood samples were taken after 14 days exposure for both nano and sulfate group. At the end of the experiment, a slight decrease trend in hematocrit, RBC, MCHC and hemoglobin observed when toxicant concentration increased in treatment tanks. In addition, Along with the decline of these parameters, MCV and MCH both increased in blood samples. Differences between nano and sulfate group was significant in 5 and 20 ppm concentration for hematocrit and RBC, and all concentrations for hemoglobin and MCHC. Furthermore, analysis of WBC showed that this parameter increased by concentration both in nano and sulfate group. Changes in WBC were significantly different between groups and zinc sulfate caused more tissue damage than nano zinc. In conclusion, this study showed that the severity of destructive effect on hematological indices were less in nano group. However, the mechanism of these changes in destructive effect is not recognized and needs further studies.

Keywords: Fish; Nanometals; Toxic Concentration; Hematology

*Correspondence to:

Mohammad Hasan Gerami, Young Researchers and Elite Club, Shiraz Branch, Islamic Azad University, Shiraz, Iran, Tel:+98 9122884418; E-mail: m.h.gerami@gmail.com

Introduction

Heavy metals are highly accumulated in aquatic ecosystems in recent years and considered the most important threat for aquatic animals (Mance, 2012). Understanding the effect of these toxicants on aquatic animals supports the larger ecotoxicological goal of comprehending the action of ecotoxicants on their population (Bols et al., 2001). Fish are one of the most widely distributed organisms in the aquatic environment, which faced with various discharged wastewater; especially heavy metals (Klaverkamp et al., 1984). Zinc, an essential element, is one of the most common heavy metal pollutants which is widely used in industry (Kori-Siakpere and Ubogu, 2008). In addition, today's with increasing nanotechnology; zinc oxide nanoparticles (ZnO-NPs) is one of the most abundantly used nanomaterial in consumer products and biomedical applications due to its specific properties, e.g. transparency, high isoelectric point, biocompatibility, and photocatalytic efficiency (Darroudi et al., 2013). Therefore, Large-scale production and consumption of this metal will result in discharge of the products containing nanozinc to aquatic ecosystems and agricultural lands (Chatterjee, 2008; Fabrega et al., 2011).

Zn toxicity to fishes is well known and studies revealed that high concentrations of zinc in water resources exerts adverse effect in fish such as accruing structural damage, affecting the growth development and survival, depressive effect on tissue respiration leading to death by hypoxia, changes in ventilatory and heart physiology, and adverse effect on hematological indices of fishes (Crespo et al., 1979; De Schampelaere and Janssen, 2004; Kori-Siakpere and Ubogu, 2008). Many researchers studied the ecotoxicity of nanomaterials to aquatic ecosystems (Moore, 2006; Handy et al., 2008; Klaine et al., 2008; Handy et al., 2011; Khabbazi et al., 2014a; 2014b). However, the environmental impacts of Zn nanoparticles (NPs) are, as yet, unknown and data on hematological indices from Zn-NPs in *Capoeta capoeta gracilis* (khramulya) are generally lacking. In addition, the relative hazard of hematology from nano forms of Zn compared to traditional metal salts is unknown.

C. capoeta gracilis, one of the subspecies of the genus *Capoeta*, is distributed throughout southwest, south and central Asia. This subspecies is a popular taxon in the south Caspian Sea basin of Iran (Abdoli, 2000). Although, this species not exploited commercially; but it has high ecological role in the Caspian Sea basin ecosystems and food webs (Patimar, 2008). This species suffers from habitat loss due to anthropogenic emissions of Zn in rivers and aquatic ecosystems. Therefore, the aim of this study is to determine the effects of dissolved Zn sulfate and Zn-NPs on the hematological indices of this species following waterborne exposure to these materials to test the adverse effect of this material on *C. capoeta gracilis*. Another goal of the current study is to compare the effects of Zn metal with Zn-NPs, to identify any nano-specific hematological on this species in contrast with Zn metal form.

Materials and Methods

Number of 100 live specimens of *C. capoeta gracilis* was obtained from Zarrin Gol River in Gorgan Province, Iran. Fish were transferred to the laboratory in Gorgan University of Agricultural and Natural Resource and accumulated randomly in 100 L aquariums for one week to reduce mortality rate due to handling stress.

Toxicity test

Acute toxicity test were performed according to O.E.C.D (Organisation for Economic Co-operation and Development, 1993) and Probit analysis to assess sublethal toxicity of Zn for this species. Furthermore, due to ethical statements, minimum number of 6 samples per replicate and trial (6×4×3, 72 total) with mean weight 15 ± 3.5 gr were exposed for 14 days to sublethal toxicity of 5, 20 and 200 ppm zinc sulfate and zinc nanoparticles and one control group (no toxicant) with three replicate. There were no significant differences between aquariums in water quality and the following characteristics were constant: pH: 7.56 ± 0.45 (TS1); temperature: $19 \pm 1^\circ\text{C}$; hardness: 293 ± 2.35 ppm and dissolved oxygen: 7.80 ± 1 mg L⁻¹ (DO-5510 Oxygen meter). 50% of the aquarium water siphoned for fecal removing every 12 h and Zn/ZnSo4 concentration re-dosed after each water change based on half of toxicant stock in each treatment tank. The photoperiod was 12 h light and 12 h dark. Zn Nano colloid was purchased from Nonaka Company, Iran (Antimicrobial Product 2 brand, mean particle size of 20-30 nm, **Figure 1**) and the colloid solved into the water by using ultrasound machine. In addition, zinc sulfate powder was purchased from SIGMA-ALDRICH Company (ZnS, Density 4.1 g/mL).

4 samples per trial and replicate (total 48) were used for blood sampling process. After 14 days experiment, blood samples were collected by cutting caudal fin and using capillary tube and stored in ice. EDTA used to prevent blood clotting. Afterwards, blood tubes were centrifuged for 5 minutes for blood plasma fractionation. Hematological parameters were estimated according to routine clinical methods (Wintrobe, 1978). The acid-hematin method of Sahli in hemometer was used to analyze hemoglobin percentage

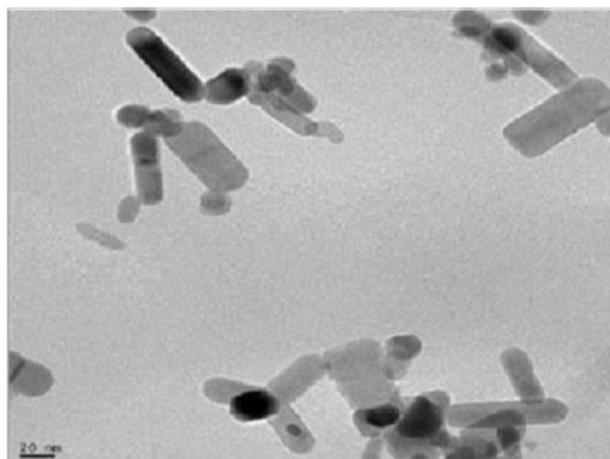


Figure 1: Particle size of Zn nanoparticles used in this study (The scale is 2.0 nm).

and Naeubaur's double hemocytometer was used to enumerate the erythrocytes (Mukherjee, 1988). Mean corpuscular volume (MCV), mean cell hemoglobin (MCH), and mean cell hemoglobin concentration (MCHC) were calculated based on Decie and Lewis (Decie and Lewis, 1991). One-way analyses of variance (ANOVA) were used to analyze hematological parameters. Differences between means were determined using Duncan's multiple range tests at 5% probability level.

Results

No mortality observed during the accumulation test and control group. Results showed a decreasing trend in hematocrit, RBC, MCHC and hemoglobin with increasing in toxicant concentration while MCV and MCH both increased by concentration. Differences between nano and sulfate group was significant in 5 and 20 ppm concentration for hematocrit and RBC, and all concentrations for hemoglobin and MCHC. In addition, analysis of WBC showed that this parameter increased by toxicant concentration both in nano and sulfate group and was significantly difference between groups (**Figure 2**). Difference between other hematological parameters in nano Zn and Zn sulfate were represented in **Figure 2**.

Discussions

Changes in the quantitative and qualitative characteristics of blood cells occur when anomalies in blood components interfere with normal functions (Landis and Yu, 2004). Results of this study showed that both form of Zn in this study, had destructive effect on hematological parameters of *C. capoeta gracilis*.

All metal exposures induced changes in fish blood indicating stress (Witeska and Kosciuk, 2003). In this study, hematocrit and hemoglobin decreased after exposure to Zn. These reduction due to Zn toxicity were also observed by Annune et al. (1994) and Kori-Siakpere and Ubogu (2008) for *Heteroclaris* species. Both hemoglobin and hematocrit are associated with RBC variation in fish blood. Haemoconcentration and haemodilution were observed in this study for both Zn form. Reduction of RBC after exposure to heavy metals was reported by Kori-Siakpere and Ubogu (2008) for *Heteroclaris* sp. and Neumosok and Hughes-24 for *Colis fasciatus* after exposure to sublethal toxicity of zinc. Witeska and Kosciuk (2003) stated that heavy metals might alter the properties of hemoglobin by decreasing their affinity towards oxygen binding capacity rendering the erythrocytes more fragile and permeable. This process could deform cell swelling and damage erythrocytes. In addition, the perturbation RBCs may be attributed to a defense reaction against toxicity through the stimulation of erythropoiesis (Maheswaran, 2008).

In the values obtained in the hematological indices, MCV increased and MCH decreased significantly after exposure to zinc (**Figure 2**). A similar observation was made for *Cyprinus carpio* after cadmium exposure (Koyama and Ozaki, 1984) and *Heteroclaris* sp. after exposure to zinc. Adeyemo (2005) stated that hypoxia or microcytic anaemia after exposure to toxicant could cause red blood cell shrinking and MCV fluctuation. In addition, fluctuation in the MCH clearly indicates that the concentration of hemoglobin in the red blood cells were much lower in the exposed fish than

in the control fish, thereby, depicting an anaemic condition (Adeyemo, 2005). Results showed significant decrease in the MCHC especially for sulfate group after exposure to zinc. This is probably an indication of red blood cell swelling and /or to a decrease in hemoglobin synthesis. Buckley et al. (1976) stated that prolonged reduction in hemoglobin content is deleterious to oxygen transport and any blood dyscrasia and degeneration of the erythrocytes could be ascribed as pathological conditions in fishes exposed to toxicants.

Significant increasing in WBC of *C. capoeta gracilis* after exposure to zinc were also observed by Kori-Siakpere and Ubogu (2008). Many researchers reported reduction in WBC after exposure to heavy metals in fishes (Remyla et al., 2008; Shah and Altindag, 2005). Changes in white blood cell count suggest dysfunction in hematological tissues (spleen and kidney) or certain infectious diseases (Khabbazi et al., 2014b). In fact, increasing or decreasing numbers of white blood cells are a normal reaction to a chemical such as zinc and demonstrating the effect of the immune system under toxic conditions. However, results showed significant decrease in lymphocyte especially in sulfate group (**Figure 2**). Witeska (2005) stated that cortisol secreted during stress reaction shortens the life span of lymphocytes, promotes their apoptosis and reduces their proliferation. Lymphocyte counts lower than normal (lymphopenia) can be an indicator of immune system deficiency and poisonous substance treatments can also deplete the body's supply of lymphocytes (Banaee et al., 2008). In addition, Banaee et al. (2008) stated that most infections cause neutrophilia. The degree of elevation often indicates the severity of the infection. Analysis showed that neutrophil increased significantly after long-term exposure especially for sulfate group.

Results showed that there were differences in destructive effect of zinc nanoparticles and zinc sulfate on *C. capoeta gracilis* hematological indices. **Figure 2** shows that toxicity of zinc sulfate decreased RBC more than nano form. This shows that zinc sulfate more destroyed RBC, and also altered their shape and size by effecting the structure and function of cell membrane. Hematocrit and hemoglobin are directly influenced by fluctuation of RBC. Results showed that these two indices were also more decreased in sulfate form. Furthermore, there were significant differences between number of WBCs in blood samples of nano and sulfate group. Results indicated that WBCs were more increased in sulfate group than nano group which indicated zinc sulfate had more tissue damage effect on *C. capoeta gracilis*. Data on comparison of nanometals versus metal ions are scarce but less severity of nano forms is well known (Handy et al., 2008; Shaw et al., 2012; Al-Bairuty et al., 2013).

Shaw and Handy (2011) declared that nanometals may not be as acutely toxic as dissolved metals. In fact, the mechanisms of uptake of nanometals are likely to be very different to dissolved metals. This process causes differences in severity of adverse effect of between nanometals and dissolved metals. In addition, metal ions have more ability to accumulate in tissue than nanoparticles. This phenomenon results more injury from metal ions. Shaw et al. (2012) studied on the adverse effect of copper nanoparticles and copper sulphate on rainbow trout and declared

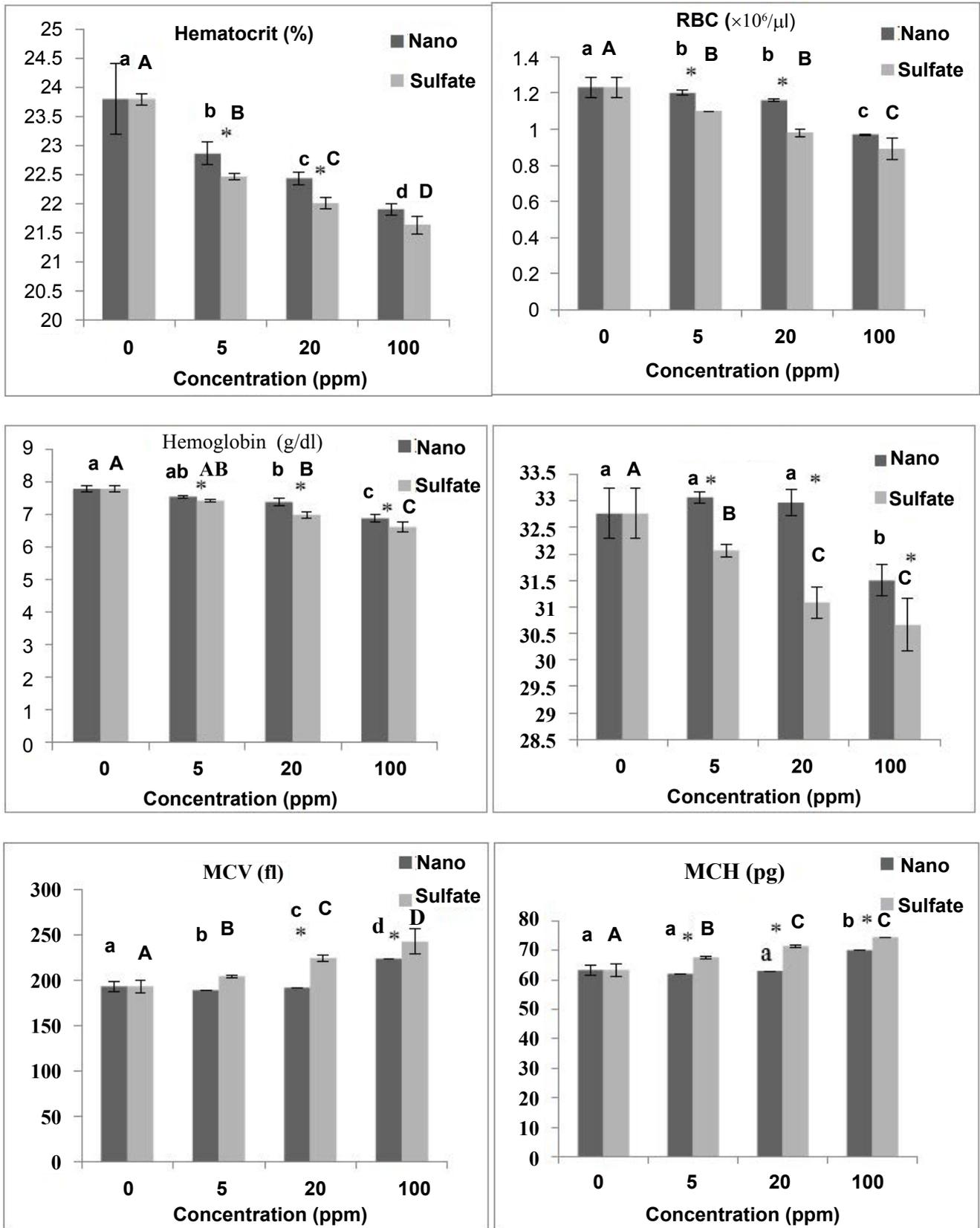


Figure 2: Hematological changes in *Capoeta capoeta gracilis* espoused to sublethal concentration of Zn nanoparticles and Zn sulfate. *Shows significant difference between nano and sulfate result for each concentration. Different capital letters and lowercase letters show significant difference between sulfate and nano result in each hematological factor, respectively (p<0.05).

that the accumulation of Cu in the gills of fish exposed to CuSO₄ is 2-fold greater than Cu nanoparticles. They also stated that while the final toxic responses may be of similar types, the mechanisms and severities driving the toxicity are different for the two forms of Cu and adverse effect of nano forms are less than the equivalent metal salts. Moreover, Shaw and Handy (2011) declared that the toxicity degree of nano and dissolved metals may be driven by the surface chemistry of the particles. These ideas are not yet proven by experimental data, but for example, one might expect a particle with an oxidising metal surface to cause oxidative stress on/in the organism. Nanoparticles have less surface area than their ion forms. These characteristic results to different solubility in water and consequently uptake level in aquatic animals. Perhaps, dissolved metals forms aggregate more than nano forms on gills and therefore, cause more stress and infectious in fishes. However, more studies need to prove this claim.

In conclusion, this study showed the degree of hematological changes in *C. capoeta gracilis* after exposure to zinc nanoparticles and zinc sulfate. Analysis showed that the severity of these changes were less in Nano group. However the mechanism of these changes in destructive effect is not recognized and needs further studies.

Ethics Statement

All experiments performed on fishes in this study complied with Society of Toxicology (code of ethics January 31, 1985; Revised June 1, 2005; Reviewed and Reaffirmed September 14, 2011; Revised November 5, 2012) and Canadian Council on Animal Care (CCAC, 1998). All analyses and experiments were performed to minimize suffering. Study was conducted with minimal number of fish.

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